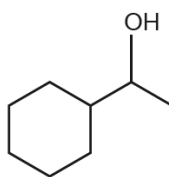


1. How can the molecule below be described?



- A Aromatic and alicyclic
- B Aliphatic and unsaturated
- C Aromatic and unsaturated
- D Alicyclic and saturated

Your answer

[1]

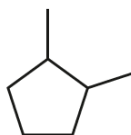
2. Which type of reaction has the greatest atom economy?

- A Substitution
- B Hydrolysis
- C Elimination
- D Addition

Your answer

[1]

3. What is the molecular formula of the compound below?

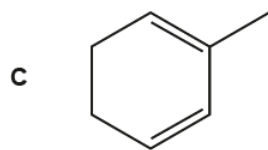
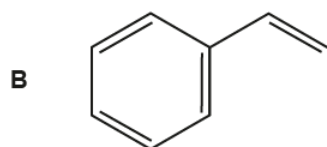
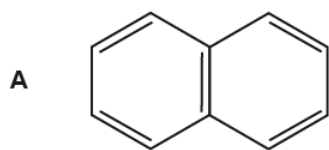


- A C_7H_{10}
- B C_7H_{12}
- C C_7H_{14}
- D C_7H_{16}

Your answer

[1]

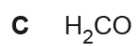
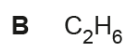
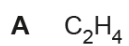
4 Which structure represents an alicyclic compound?



Your answer

[1]

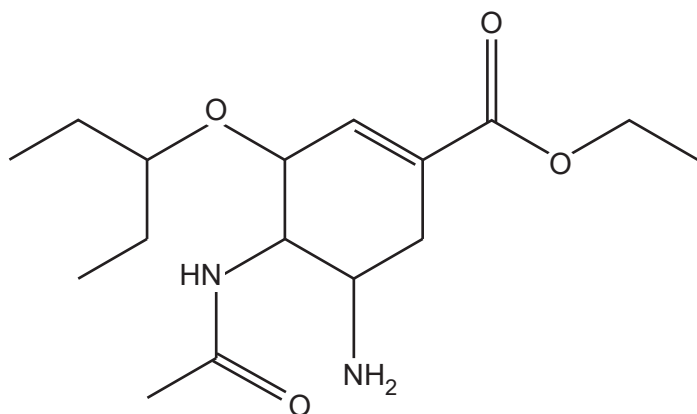
5 Which molecule is **not** planar?



Your answer

[1]

6. The structure of a compound used to treat influenza is shown below.



Which functional group(s) is/are in a molecule of the compound?

- 1 Ester
- 2 Secondary amide
- 3 Ketone

- A** 1, 2 and 3
B Only 1 and 2
C Only 2 and 3
D Only 1

Your answer

[1]

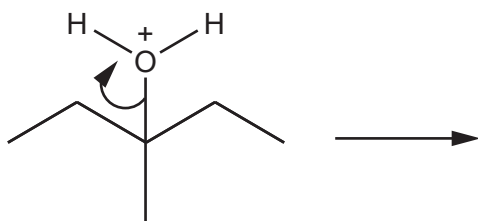
7. This question is about reaction mechanisms.

(a) Chemists use curly arrows in reaction mechanisms.

(i) What does a curly arrow show in a reaction mechanism?

.....
 [1]

(ii) Draw structures to show the products in the reaction mechanism below.



[2]

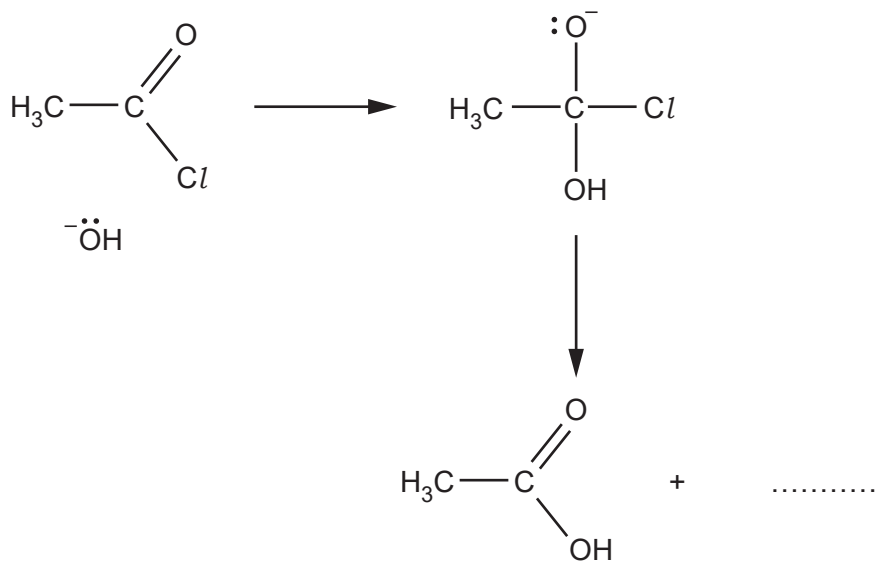
(iii) Use the mechanism in (ii) to explain what is meant by **heterolytic fission**.

.....

 [2]

(b) An incomplete reaction mechanism is shown below.

(i) Complete the mechanism by adding curly arrows and any missing species.



[4]

(ii) What is the role of OH^- in this mechanism?

..... [1]

8. What is the number of alicyclic structural isomers of C_5H_{10} ?

A 3

B 4

C 5

D 6

Your answer

[1]

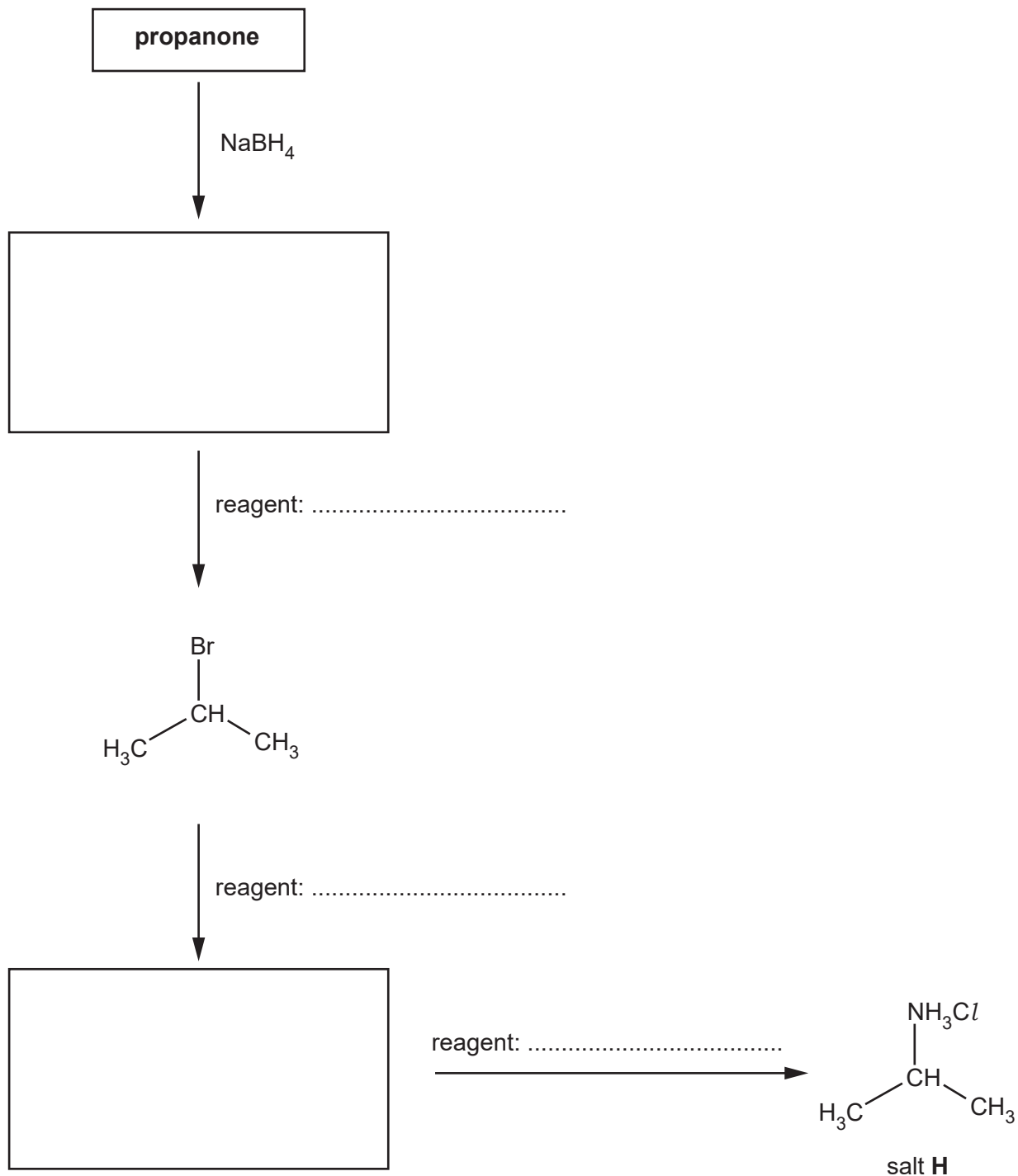
9. This question is about organic compounds containing nitrogen.

(a) Salt **H**, $(\text{CH}_3)_2\text{CHNH}_3\text{Cl}$, is used in the manufacture of garden weedkillers.

The flowchart shows the synthesis of the salt **H** from propanone.

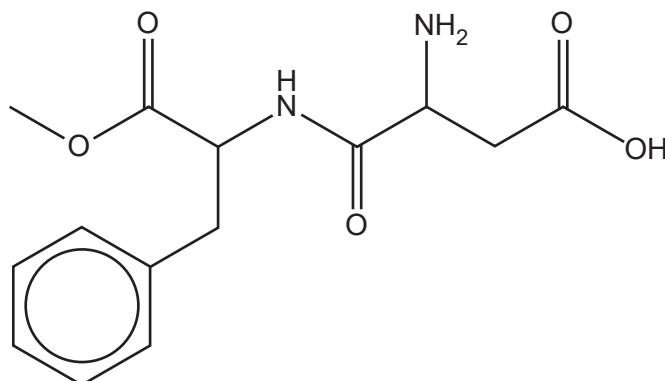
Complete the flowchart.

Show structures for organic compounds.



[5]

(b) Aspartame, shown below, is an artificial sweetener commonly used as a sugar substitute.



aspartame

(i) Aspartame contains several functional groups.

Apart from the benzene ring, name the functional groups in aspartame.

.....

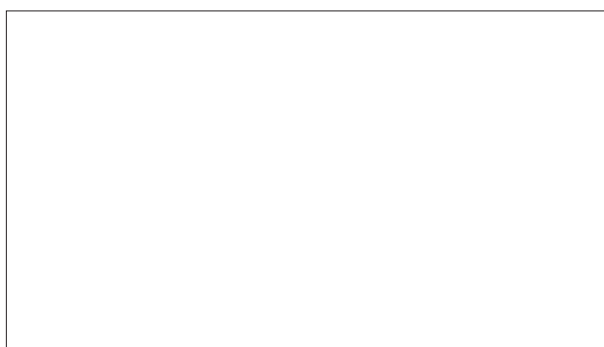
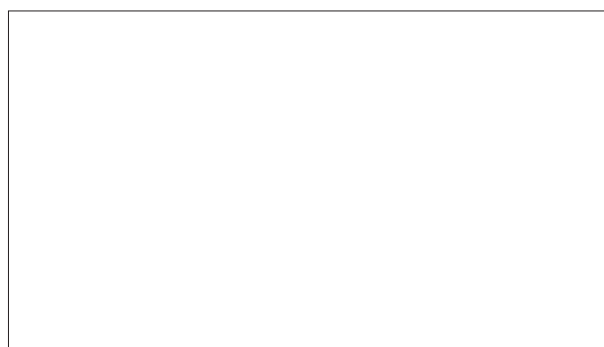
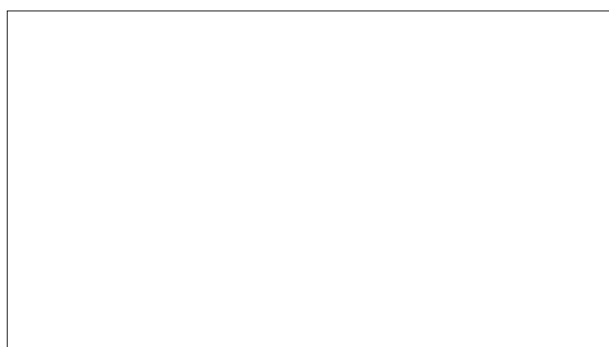
.....

.....

..... [3]

(ii) A sample of aspartame is hydrolysed with aqueous acid.

Draw the structures of the **three** organic products of the complete **acid hydrolysis** of aspartame.



[4]

(iii) Some people are concerned that aspartame, $C_{14}H_{18}N_2O_5$, may have adverse health effects.

Research shows that the safe maximum daily intake of aspartame is $1.7 \times 10^{-4} \text{ mol kg}^{-1}$.

- A typical UK adult has a mass of 75 kg.
- A can of a diet drink contains 167 mg of aspartame.

How many cans of this diet drink is it safe for a typical adult to drink in one day?

Number of cans = [3]